

## 5-Bromo-1,3-dichloro-2-iodobenzene

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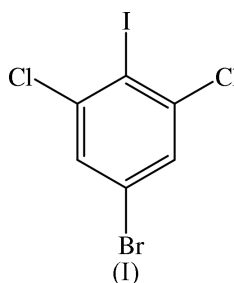
## Key indicators

Single-crystal X-ray study  
 $T = 293\text{ K}$   
Mean  $\sigma(\text{C}-\text{C}) = 0.015\text{ \AA}$   
 $R$  factor = 0.057  
 $wR$  factor = 0.147  
Data-to-parameter ratio = 17.3For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound,  $\text{C}_6\text{H}_2\text{BrCl}_2\text{I}$ , crystallizes in space group  $P2_1/c$  with two molecules in the asymmetric unit. The molecules stack in two different directions, with their plane normals approximately parallel to  $[110]$  and  $[1\bar{1}0]$ . The molecules of the asymmetric unit are held together by  $\pi$ - $\pi$  interactions.

## Comment

The title compound, (I), was synthesized for a study of the electrochemical reductive cleavage of carbon-halogen bonds (Arun Prasad & Sangaranarayanan, 2004). The average C-Cl bond length is 1.735 Å, which is in excellent agreement with the normal value of 1.739 Å (Allen *et al.*, 1987). Likewise, the average C-Br and C-I bond lengths of 1.879 and 2.086 Å are in good agreement with the normal values of 1.899 and 2.095 Å, respectively (Allen *et al.*, 1987). The halogen atoms are very slightly displaced from the benzene ring plane, as a result of steric repulsion; the largest deviation is 0.089 (3) Å for atom I2. Such steric interactions between the halogens in polyhalobenzenes are well known (Solenova *et al.*, 1960).

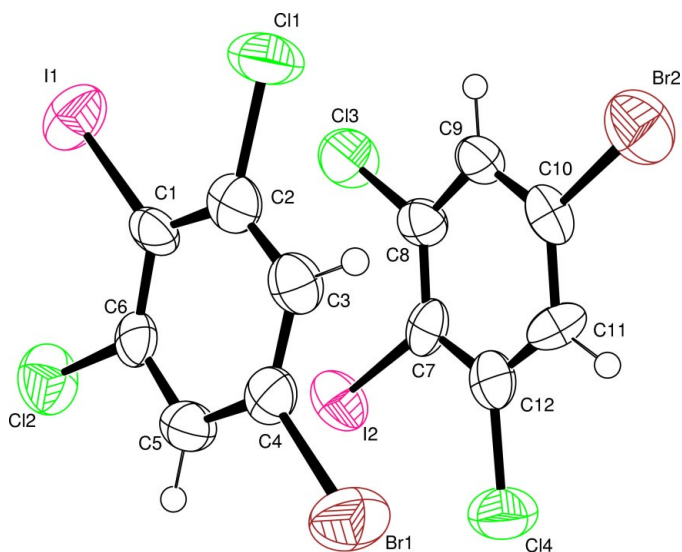


The asymmetric unit of (I) contains two molecules that are nearly parallel to each other, the angle between their plane normals being 10.82 (5)°. The approximate perpendicular distance between the planes, 3.42 Å, is indicative of  $\pi$ - $\pi$  interactions. The asymmetric unit pairs stack along planes normal to  $[110]$  and  $[1\bar{1}0]$ . The perpendicular distance of 3.87 Å between the stacked pairs suggests that there is no extended  $\pi$ - $\pi$  interaction in the stacking direction.

## Experimental

2,6-Dichloroaniline was brominated by passing bromine vapour into a solution of 2,6-dichloroaniline (21 g) in hydrochloric acid (6 M, 120 ml). Solid 4-bromo-2,6-dichloroaniline was filtered off and purified by column chromatography (silica gel). 4-Bromo-2,6-dichloroaniline (30 g) was then diazotized in hydrochloric acid (6 M, 150 ml) using aqueous sodium nitrite (15 g/36 ml water) and the resulting solution was slowly added to aqueous potassium iodide (36 g/45 ml

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**Figure 1**  
The asymmetric unit of the title compound, with the atomic numbering scheme. Displacement parameter ellipsoids are drawn at the 50% probability level.

water). When no further gas was evolved, the crude product was filtered off, and washed first with aqueous sodium hydroxide, subsequently with sodium metabisulphite and finally with water. 5-Bromo-1,3-dichloro-2-iodobenzene was separated and purified by column chromatography (silica gel) using hexane as the eluant; slow solvent evaporation produced X-ray quality crystals of (I).

#### Crystal data

$C_6H_2BrCl_2I$   
 $M_r = 351.79$   
Monoclinic,  $P2_1/c$   
 $a = 11.630$  (6) Å  
 $b = 10.020$  (4) Å  
 $c = 16.30$  (1) Å  
 $\beta = 109.84$  (4)°  
 $V = 1786.6$  (15) Å<sup>3</sup>  
 $Z = 8$

$D_x = 2.616$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
Cell parameters from 25 reflections  
 $\theta = 10$ – $15^\circ$   
 $\mu = 8.58$  mm<sup>-1</sup>  
 $T = 293$  (2) K  
Block, colourless  
 $0.3 \times 0.2 \times 0.2$  mm

#### Data collection

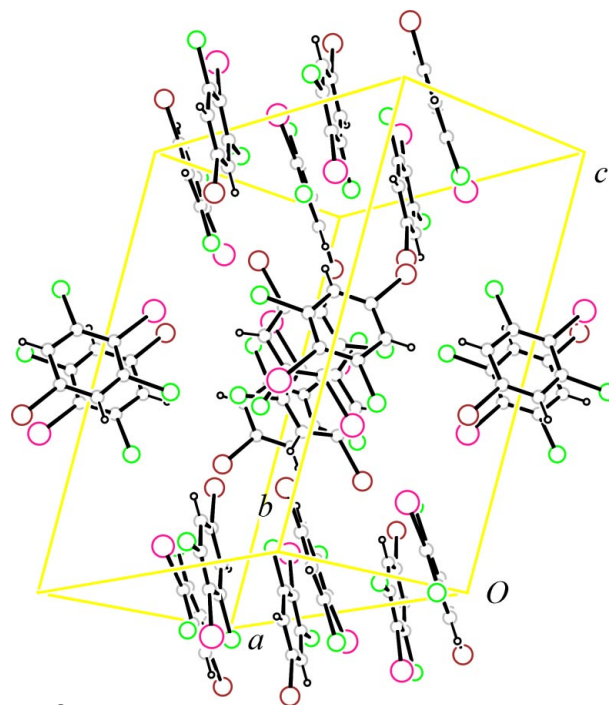
Enraf–Nonius CAD-4 diffractometer  
 $\omega/2\theta$  scans  
Absorption correction:  $\psi$  scan (North *et al.*, 1968)  
 $T_{\min} = 0.361$ ,  $T_{\max} = 1.000$   
3287 measured reflections  
3125 independent reflections  
1730 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.025$   
 $\theta_{\text{max}} = 25.0^\circ$   
 $h = 0 \rightarrow 13$   
 $k = 0 \rightarrow 11$   
 $l = -19 \rightarrow 18$   
2 standard reflections  
frequency: 60 min  
intensity decay: none

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.057$   
 $wR(F^2) = 0.147$   
 $S = 0.92$   
3125 reflections  
181 parameters

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0856P)^2]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 1.16$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -1.71$  e Å<sup>-3</sup>



**Figure 2**  
A packing diagram of the title compound, viewed down  $[110]$ .

All H atoms were placed in calculated positions, with C–H = 0.93 Å, and were refined using the riding-model approximation, with  $U_{\text{iso}}$  values constrained to  $1.2U_{\text{eq}}$  of the carrier atom. The largest residual electron-density peak is located 1.13 Å from atom I1 and the deepest hole is located 1.20 Å from atom I1.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1989); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* in *WinGX* (Farrugia, 1999); program(s) used to solve structure: *SIR92* in *WinGX*; program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP3* in *WinGX*; software used to prepare material for publication: *SHELXL97*.

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